

ISOLATION AND STRUCTURE  
OF OXIRAPENTYN

Sir:

A new antibiotic, oxirapentyn was isolated from the culture broth of fungus identified as *Beauveria felina* SANK 13682. It was characterized to have a new skeleton with two isoprene units and an activity against Gram-positive bacteria. In this paper we report the isolation, biological activities and structure of oxirapentyn.

Oxirapentyn was produced in a shaken culture of *B. felina* SANK 13682 for 3 to 5 days at 26°C in a medium containing 2% sucrose, 2.5% mashed potatoes, 1% Casamino Acids, 0.5%  $\text{KH}_2\text{PO}_4$  and 0.25%  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ . The culture filtrate (8.7 liters) was extracted with ethyl acetate (9 liters) and the solvent layer was concentrated *in vacuo* to syrup. This syrup was dissolved in 10 ml of benzene - ethyl acetate (9:1) and further purified by silica gel column chromatography. The active fractions were concentrated to yield the crude oxirapentyn. The pure oxirapentyn (1.3 g) was obtained by recrystallization from the solvent

mixture of hexane - ethyl acetate as colorless needles.

Oxirapentyn (**1**): soluble in methanol, acetone, ethyl acetate, and chloroform and insoluble in water; mp 114~115°C; *Anal.* calcd. for  $\text{C}_{15}\text{H}_{20}\text{O}_6$ : C 65.05, H 6.07, found: C 65.42, H 6.05; MS  $m/z$  332 ( $\text{M}^+$ ); UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $E_{1\text{cm}}^{1\%}$ ) 225 (480);  $[\alpha]_{\text{D}}^{20}$   $-111.7^\circ$  ( $c$  1,  $\text{CHCl}_3$ ); IR  $\nu_{\text{max}}^{\text{KBr}}$  2230 (triple bond), 1740 (acetyl), 1710 (ketone) and 1608  $\text{cm}^{-1}$  (double bond). The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of **1** are shown in Figs. 1 and 2. The complete structure of **1** was determined by  $J_{\text{CC}}$  coupling and the long-range  $J_{\text{CH}}$  resolved 2-D method in the NMR spectrum, the result of which will be discussed elsewhere. However, the structure of **1** with its relative configuration was elucidated by means of X-ray analysis as follows: The crystals are trigonal, space group  $\text{P3}_121$  (or  $\text{P3}_221$ ) with  $a=11.078$  (2),  $c=25.170$  (4) Å,  $Z=6$ . Intensity data were measured on a Rigaku AFC-5 apparatus equipped with a rotating-anode X-ray generator (graphite-monochromated  $\text{Cu-K}\alpha$  radiation). The structure was solved by MULTAN<sup>1)</sup> and refined by block-diagonal least-

Fig. 1.  $^1\text{H}$  NMR spectrum of oxirapentyn (90 MHz,  $\text{CDCl}_3$ ).

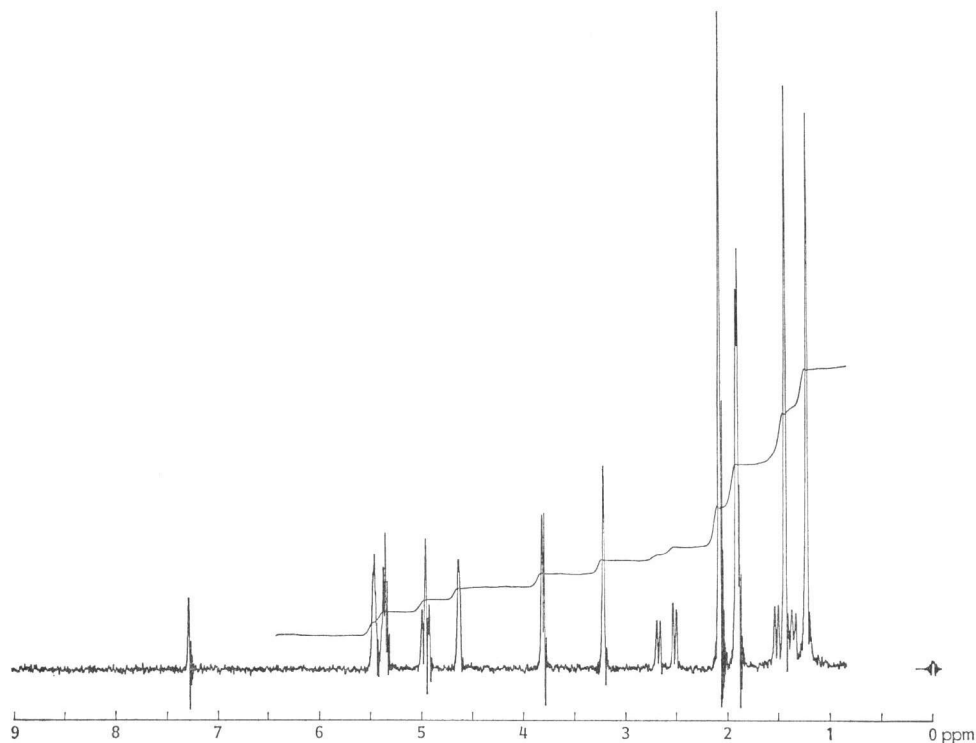


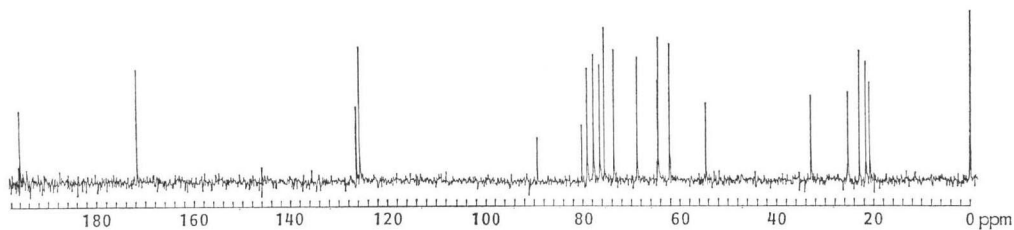
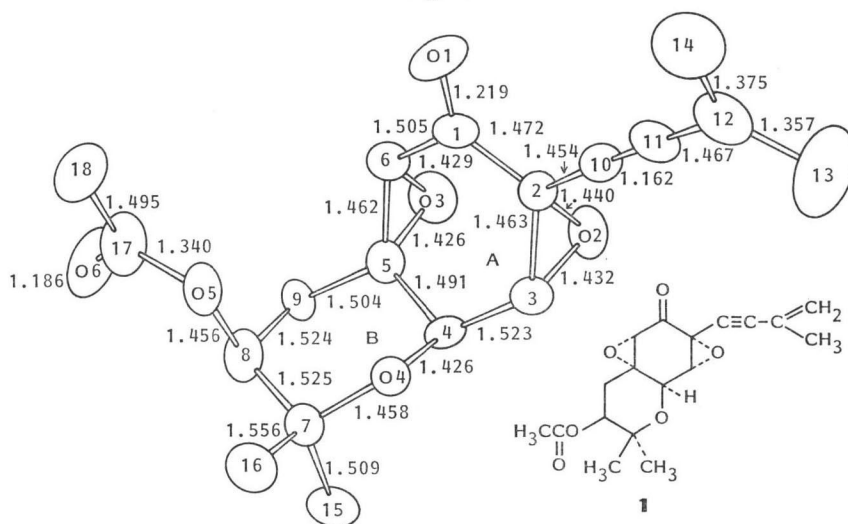
Fig. 2.  $^{13}\text{C}$  NMR spectrum of oxirapentyn ( $\text{CDCl}_3$ ).

Fig. 3.



squares methods. Hydrogen atoms were located from a difference Fourier synthesis. The final least-squares refinement with anisotropic temperature factors for the non-hydrogen atoms and isotropic temperature factors for the hydrogen atoms lowered the R value of 6.7% for 1,020 observed reflections ( $F_o \geq 1.5 \sigma F_o$ ). The final atomic parameters have been deposited with the Crystallographic Data Center. The molecular skeleton thus obtained is shown in Fig. 3 drawn by ORTEP<sup>2)</sup>. The two six-membered A and B rings are *cis*-fused, and the A ring takes a flattened boat form with C(1) as a top and C(4), an end atom, while the B ring takes a chair form with O(4) as a stem and C(9), a stern atom. The two epoxy groups attached to the A ring are *syn* with respect to each other. The *O*-acetyl group attached to the B ring is in an  $\beta$ -axial position. The relatively high temperature parameters (C(13)=8.8 and C(14)=11.6  $\text{\AA}^2$ ) of the isopropenyl carbon atoms may reflect a large thermal motion and/or the disordering of atoms. Thus, the bond

Table 1. Antimicrobial spectrum of oxirapentyn.

Test organism	Medium*	MIC ( $\mu\text{g/ml}$ )
<i>Staphylococcus aureus</i> FDA 209P JC-1	1	25
<i>Streptococcus faecalis</i> S-299	1	800
<i>Bacillus subtilis</i> PCI 219	1	25
<i>Micrococcus luteus</i> PCI 1001	1	25
<i>Mycobacterium smegmatis</i> ATCC 607	1	100
<i>Escherichia coli</i> NIHJ JC-2	1	>800
<i>Klebsiella pneumoniae</i> PCI 602	1	800
<i>Proteus vulgaris</i> OX19	1	>800
<i>Pseudomonas aeruginosa</i> NCTC 10490	1	>800
<i>Bacteroides fragilis</i> SANK 71176	2	800
<i>Eubacterium aerofaciens</i> SANK 72276	2	>800
<i>Propionibacterium acnes</i> SANK 71976	2	400
<i>S. faecalis</i> S-299	2	>800

\* 1. Mueller Hinton agar.

2. GAM agar, under anaerobic condition.

order of the C(12)-C(13) and C(12)-C(14) bonds could not be given. The absolute configuration of **1** is still ambiguous and under investigation by X-ray analysis and CD spectrum.

The antimicrobial activity of **1** determined by the agar dilution method is shown in Table 1. The LD<sub>50</sub> of **1** to mice injected intraperitoneally was 6.25 mg/kg.

SHUJI TAKAHASHI\*  
YASUHIRO ITOH  
MICHIKO TAKEUCHI  
KOUHEI FURUYA  
KENTARO KODAMA  
ATSUSHI NAITO  
TATSUO HANEISHI  
SADAO SATO\*\*  
CHIHIRO TAMURA\*\*

Fermentation Research  
Laboratories,  
\*\*Analytical and Metabolic  
Research Laboratories,  
Sankyo Co. Ltd.  
1-2-58, Hiromachi, Shinagawa-ku,  
Tokyo, 140, Japan

(Received June 22, 1983)

#### References

- 1) GERMAIN, G.; P. MAIN & H. W. WOOLFSON: The application of phase relationships to complex structures. III. The optimum use of phase relationships. *Acta Cryst.* A27: 368~376, 1971
- 2) JOHNSON, C. K.: "ORTEP, Oak Ridge National Laboratory Report ORNL-3794" 1965